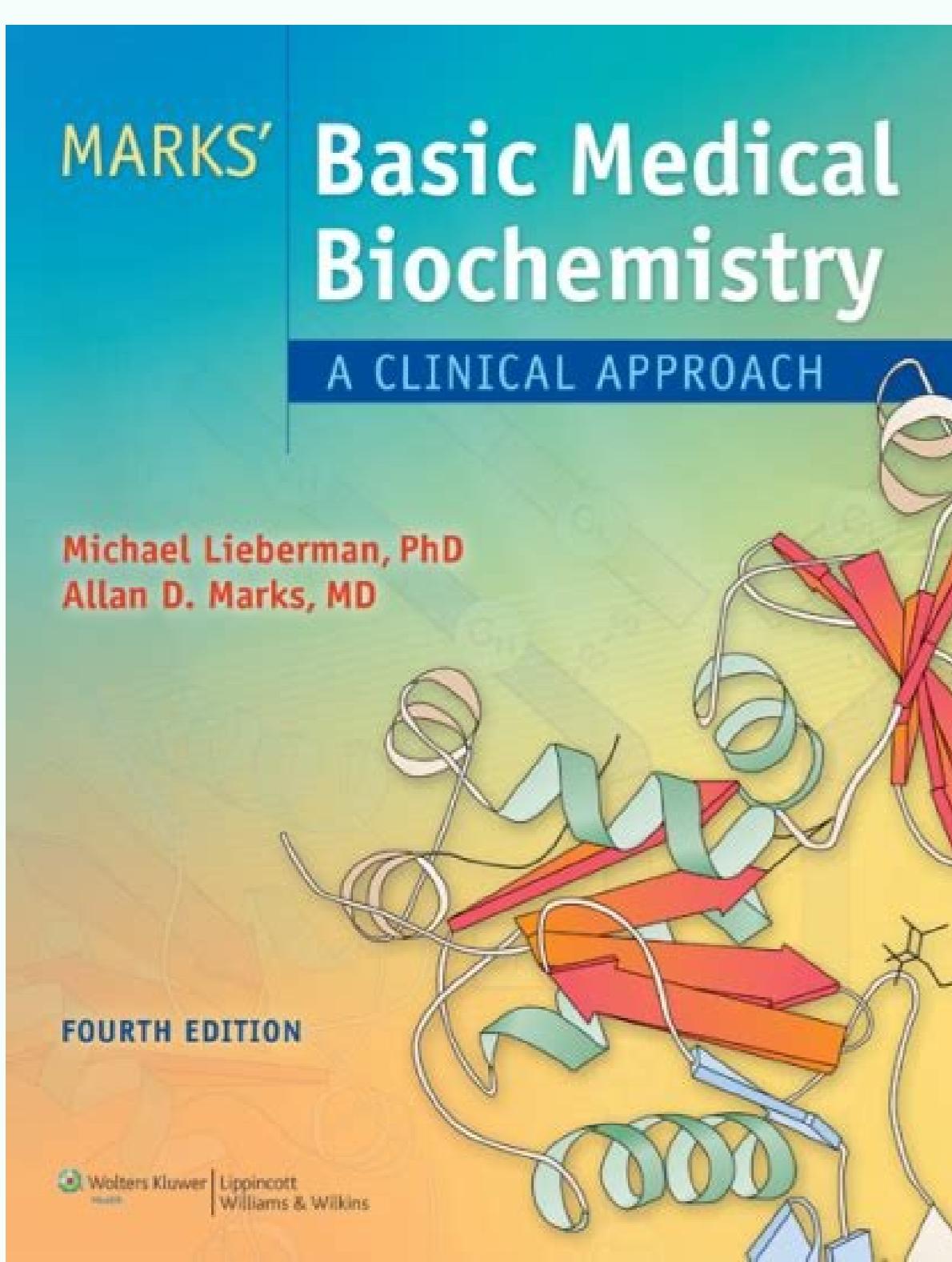
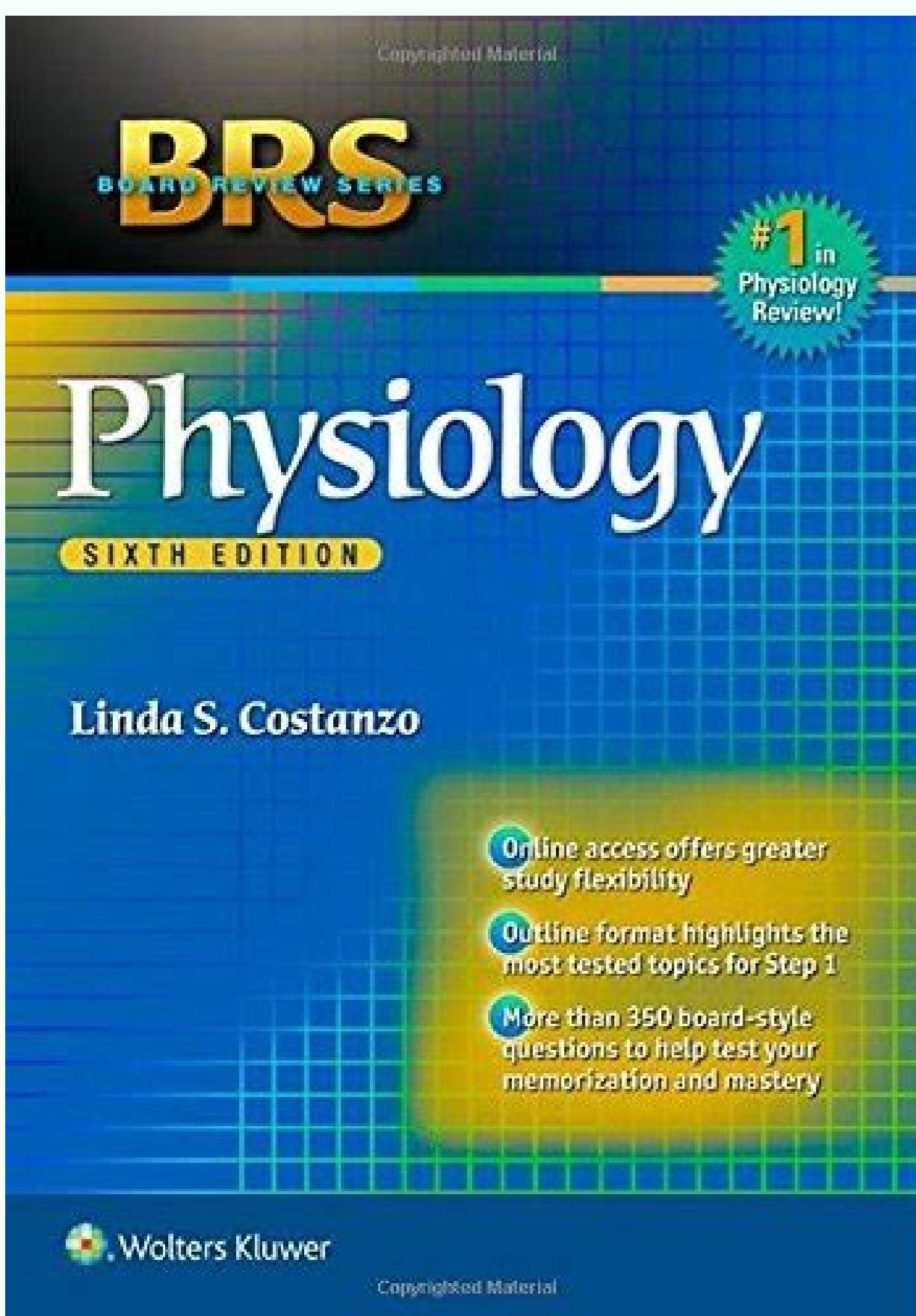


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Biology MCQ (Multiple Choice Questions in Life Science)

{Sample/Model/Practice Questions for JRF/NET Life Science Examination, ICMR/JRF, DBT/JRF, GATE, IIT-JAM, IIT-JEE, PG Entrance}

MCQ Biochemistry: Amino Acids Part 4 (MCQ 008)

- Which group of a fully protonated glycine ($\text{NH}_3^+ - \text{CH}_2 - \text{COOH}$) first release a 'proton' when it is titrated against -OH^- ions?
 - Carboxyl group
 - Amino group
 - Both at the same time
 - It cannot be predicted
 - pK_a is the measure of a group to _____ proton.
 - Take up
 - Release
 - Combine
 - Consume
 - Which of the following amino acid bears a guanidine group in the side chain?
 - Lysine
 - Arginine
 - Histidine
 - Proline
 - The precursor of glycine synthesis in microbes and plants is _____.
 - Serine
 - Leucine
 - Valine
 - None of these
 - Single letter code of selenocysteine is _____.
 - B
 - J
 - U
 - O
 - Which of the following amino acid have an imino group in the side chain?
 - Proline
 - Asparagine
 - Glutamate
 - Histidine
 - 4-hydroxy proline (a derivative of proline) is abundantly present in _____.
 - Keratin
 - Myoglobin
 - Hemoglobin
 - Collagen
 - Desmosine is a complex derivative of five _____ residues.
 - Lysine
 - Arginine
 - Histidine
 - Methionine
 - Isoelectric pH is designated as _____.
 - pKa
 - pI
 - pH

For more Questions and Resources for your CSIR/ICMR/DBT/ICAR/UGC/JRF NET Life Science Examination please visit:

www.easybiologyclass.com

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Access to all 3 MedSchoolCoach Full-Length Practice Exams (\$110 value) Access to our MCAT Prep App, containing 200+ videos & lecture notes, 1000+ flashcards and questions for MCAT practice 6 Months of Access to MCAT Go - Our complete science audio course (\$600 value) To help pre-med students with their MCAT prep, the Association of American Medical Colleges (AAMC) provides some free resources, including an MCAT Sample Question Guide that contains 12 MCAT questions (three questions from each of the four sections tested on the MCAT). Additionally, students can find 100 free MCAT practice questions with their free seven-day trial ... The question asks us to find the pH of the solution, so we will need to convert pOH to pH. To do so, we simply subtract the pOH from 14. The pH of the solution is 12.3. Because sodium hydroxide is a strong base, it makes sense that the pH is above 7. Example Question #9 : Impulse And Momentum Joe, of mass 90kg, jumps straight up. To do so, he bends his knees and produces an upwards force that ...

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